Toward a Unique Definition of the Local Spin

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ABSTRACT: In this work, we demonstrate that there is a continuum of different formulations for the decomposition of $\langle S^2 \rangle$ that fulfills all physical requirements imposed to date. We introduce a new criterion based upon the behavior of single-electron systems to fix the value of the parameter defining that continuum, and thus we put forward a new general formula applicable for both single-determinant and correlated wave functions. The numerical implementation has been carried out in the three-dimensional physical space for several atomic definitions. A series of representative closed-shell and open-shell systems have been used to illustrate the performance of this new decomposition scheme against other existing approaches. Unlike other decompositions of $\langle S^2 \rangle$, the new scheme provides very small local-spin values for genuine diamagnetic molecules treated with correlated wave functions, in conformity with the physical expectations.

1. INTRODUCTION

The spin state of an atom or a group of atoms within a molecule is not a quantum mechanical observable. Nevertheless, the concept of local spin is generally invoked when discussing spin–spin interactions between magnetic centers in a molecule with Heisenberg Hamiltonian models or to describe organic molecules with diradical character. For instance, the spin on transition metal or diatomic complexes is particularly relevant for experimentalists and can be used for spin-labeling functional groups. In addition, local-spin values could also be used as an stringent test for DFT and DMFT functionals in order to analyze their ability to recover the spin distribution along the molecular space.

In recent years, there has been a growing interest in recovering local spins from ab initio wave functions.1−14 These techniques aim to express the expectation value of the $\hat{S}^2$ operator as a sum of atomic and diatomic contributions:

$$\langle \hat{S}^2 \rangle = \sum_A \langle S^2 \rangle_A + \sum_{A,B \neq A} \langle S^2 \rangle_{AB} \quad (1)$$

In 2001, Clark and Davidson1 proposed a general framework based on the definition of local (atomic) spin operators, $\hat{S}_A$, obtained by proper projections of the overall spin vector operator. The expectation values of the atomic spin-square operators and diatomic products of spin operators, namely $\langle \hat{S}_A^2 \rangle$ and $\langle \hat{S}_A \hat{S}_B \rangle$, can be considered as the atomic and diatomic contributions to the overall $\langle \hat{S}^2 \rangle$ value, respectively. Unfortunately, this formalism leads to significant local-spin contributions for genuine closed-shell diamagnetic molecular systems such as the H₂ molecule treated at the RHF level of theory.

Later on, Mayer4 proposed an alternative strategy to partition the expectation value of the $\hat{S}^2$ operator in the spirit of classical population analysis. In the case of single-determinant wave functions, Mayer obtained expressions for the atomic and diatomic contributions to $\langle \hat{S}^2 \rangle$ depending only on the molecular spin-density matrix, $\rho^S (\tilde{r} \tilde{r})$. Such a decomposition trivially leads to zero atomic spins for any closed-shell restricted wave function.

This decomposition was carried out originally in the framework of the so-called Hilbert-space analysis. The generalization of the formulas to the 3D physical-space analysis was already outlined by Mayer4 and has also been very recently explored by Alcoba et al.,5 using both Bader’s QTAIM15 domains and Becke’s “fuzzy” atoms.16

Many molecular species do not admit single-determinant description. In some singlet systems such as diradicals or in the dissociation process of a molecule, where one can also recognize the existence of local spins, a proper spin description can only be accomplished using multideterminant wave functions, whose $\langle \hat{S}^2 \rangle$ decomposition is not straightforward. Indeed, in recent years, a number of different $\langle \hat{S}^2 \rangle$ decomposition schemes, exhibiting different features, have been proposed.5−7,9 In the first scheme, proposed by Alcoba et al.,5 the components of $\langle \hat{S}^2 \rangle$ were expressed in terms of the total spin-density matrix, and therefore zero spins were obtained for every singlet state system. Subsequently, Mayer6 proposed an alternative formulation on the basis of several physical requirements: (i) to obtain zero local spins for closed-shell restricted wave functions; (ii) proper asymptotics, i.e., in the dissociation limit the atoms/fragments should have the same local $\langle \hat{S}^2 \rangle$ value as the respective free atoms/fragment radicals; (iii) the formula used for a general wave function should reduce to that used in the single-determinant case if applied to single-determinant wave functions.

In this formulation, the atomic and diatomic components of $\langle \hat{S}^2 \rangle$ were expressed in terms of the value that they would have in the single-determinant case, plus correction terms arising from the deviation of the first-order density matrix from the indempotency, as well as contributions from the cumulant of the second-order density matrix. (Both vanish for single determinant wave functions, thus fulfilling requirement iii.) These correction terms were distributed between atomic and diatomic contributions in such a way that the local spins obtained for atoms or other fragments at large distances coincide with those of the free atoms. The Hilbert-space realization of

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the formalism given in ref 7 corroborated the physical expectations in all cases. For instance, small atomic contributions were obtained for every closed-shell molecule at equilibrium distance, including, e.g., the carbon atoms of singlet π-conjugated systems such as benzene calculated at the CASSCF level of theory. At the same time, large local spins were found for the square cyclo-butadiene, which is known to be a "molecular antiferromagnet". Furthermore, the CASSCF dissociation curves of the O2 molecule for both singlet and triplet states lead to atomic local-spin values that tend asymptotically to 2, as expected for the 1P2 state of the free oxygen atom. Similarly, ethylene dissociates into two triplet methylene radicals, and so forth.

Although it seemed as there was no further freedom to choose another local-spin decomposition scheme, Alcoba et al.9 have recently shown that this is not the case. These authors have introduced yet another requirement for the partitioning of \( \langle S^2 \rangle \) for nonsinglet states, according to which the one- and two-center components should be independent of the \( S_r \) value (in the absence of magnetic field). Their partitioning scheme uses "spin-free" quantities such as the effectively unpaired density matrix17 and the "spin-free cumulant matrix", which depend on the spin-free first- and second-order reduced density matrices.18,19 Their Hilbert-space local-spin results are similar to but not identical with those obtained by Mayer and Matito in ref 7 and keep the physical requirement of \( S_r \) independence.

In this paper, we shall show that there is actually a continuum of different formulations that fulfill all physical requirements imposed to date to the decomposition of \( \langle S^2 \rangle \), and this applies to both single- and multisdeterminant wave functions. In order to find the best local-spin decomposition scheme, we impose a new, additional requirement related to the one-electron distribution that eliminates the ambiguity. The resulting local-spin decomposition is thus unique and fulfills all physical requirements found so far.

The paper is organized as follows: first, we put forward the definition against previous definitions. Then, we suggest a unique definition of the local-spin decomposition scheme, Alcoba et al.9 by a general wave function can be expressed in terms of the spinless second-order density matrix as (see Appendix A1 for further details)

\[
\langle S^2 \rangle = \frac{3}{4} \int \rho(\vec{r}_1; \vec{r}_1) \, d\vec{r}_1 - \frac{1}{4} \int \rho(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \, d\vec{r}_1 \, d\vec{r}_2 - \frac{1}{2} \int \rho(\vec{r}_1; \vec{r}_2; \vec{r}_1) \, d\vec{r}_1 \, d\vec{r}_2 \tag{3}
\]

where the first- and second-order density matrices are normalized to \( N \) and \( N(N-1) \) (note that this normalization is different from that used by Alcoba et al.), respectively. The spinless second-order density matrix is often expressed in terms of the first-order density matrix elements and the so-called cumulant of the second-order density matrix, \( \Gamma(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \) (which vanishes for single-determinant wave functions) as

\[
\rho_2(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) = \rho(\vec{r}_1; \vec{r}_2) + \frac{1}{2} \rho(\vec{r}_1; \vec{r}_2; \vec{r}_1) + \frac{1}{2} \rho(\vec{r}_1; \vec{r}_2; \vec{r}_1) + \frac{1}{2} \rho(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \tag{4}
\]

where \( \rho(\vec{r}; \vec{r}) \) is the first-order spin-density matrix, defined as the difference between the \( \alpha \) and \( \beta \) components of the first-order density matrix

\[
\rho^\alpha(\vec{r}; \vec{r}) = \rho^\beta(\vec{r}; \vec{r}) - \rho^\beta(\vec{r}; \vec{r}) \tag{5}
\]

The spinless cumulant \( \Gamma(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \) represents the sum of the four respective spin-dependent cumulants and differs from the entity \( \langle S^2 \rangle \) that Alcoba et al. called "spin-free cumulant". Substituting eq 4 into eq 3, one gets

\[
\langle S^2 \rangle = \frac{3}{4} \int \rho(\vec{r}_1) \, d\vec{r}_1 - \frac{3}{8} \int \rho(\vec{r}_1; \vec{r}_2) \rho(\vec{r}_1; \vec{r}_1) \, d\vec{r}_1 \, d\vec{r}_2 + \frac{1}{8} \int \rho^\alpha(\vec{r}_1; \vec{r}_2) \rho^\beta(\vec{r}_1; \vec{r}_1) \, d\vec{r}_1 \, d\vec{r}_2 + \frac{1}{4} \int \rho(\vec{r}_1) \, d\vec{r}_1 \int \rho(\vec{r}_2) \, d\vec{r}_2 - \frac{1}{4} \int \Gamma(\vec{r}_1; \vec{r}_2) \, d\vec{r}_1 \, d\vec{r}_2 - \frac{1}{2} \int \Gamma(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \, d\vec{r}_1 \, d\vec{r}_2 \tag{6}
\]

where \( \rho(\vec{r}_1; \vec{r}_2) \) and \( \Gamma(\vec{r}_1; \vec{r}_2; \vec{r}_1; \vec{r}_2) \) vanish. In fact, the natural decomposition of eq 6 leads to restricted single-determinant wave functions, and therefore, for such wave functions, the decomposition of eq 8 leads to identically zero spin components, thus fulfilling requirement 1. This is not the case for expression 6, since the first and second terms on the rhs can be decomposed into nonzero one-center and one- and two-center contributions, respectively, although their sum vanishes. In fact, the natural decomposition of eq 6 into one- and two-center terms leads to Clark and Davidson's formulas.20 Moreover, Alcoba et al.9 pointed out that since \( \rho(\vec{r}_1; \vec{r}_2) \) and
\(\rho_s(\vec{r}_1, \vec{r}_2)\) are spin-independent quantities, by virtue of eq 4, the sum
\[
\Gamma(\vec{r}_1, \vec{r}_2; \vec{r}_1', \vec{r}_2') - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1')
\]

must be spin-independent too (it is essentially their “spin-free cumulant”). Also, they showed that (in the normalization used)
\[
\int \left[ \Gamma(\vec{r}_1, \vec{r}_2) - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1) \right] \, d\vec{r}_2 = -\frac{1}{2} u(\vec{r}_1)
\]  

On this basis, they suggested the following general expression for \(\langle \vec{S}^2 \rangle\)
\[
\langle \vec{S}^2 \rangle = \frac{1}{2} \int u(\vec{r}_1) \, d\vec{r}_1 - \frac{1}{2} \int \left[ \Gamma(\vec{r}_1, \vec{r}_2; \vec{r}_1', \vec{r}_2') - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1') \right] \, d\vec{r}_1 \, d\vec{r}_2
\]

Both expression 8 and Alcoba’s 11 are natural starting points to derive \(\langle \vec{S}^2 \rangle\) components that fulfill the imposed requirement to be independent of the \(S_z\) projection. Upon integration over the whole space, both expressions are equivalent. However, the calculation of the one- and two-electron integrals over one- and two-center atomic domains leads to different contributions. In fact, one can freely modulate the relative weight of the first two terms on the rhs of eq 8 with a parameter \(a\) and get a general expression like
\[
\langle \vec{S}^2 \rangle = a \int u(\vec{r}_1) \, d\vec{r}_1 - (1 - 2a) \int \left[ \Gamma(\vec{r}_1, \vec{r}_2; \vec{r}_1', \vec{r}_2') - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1') \right] \, d\vec{r}_1 \, d\vec{r}_2
\]  

which leads to different decompositions of \(\langle \vec{S}^2 \rangle\) that fulfill requirements i–iii. (The two equations discussed above correspond to the values \(a = 3/8\) and \(a = 1/2\), respectively.) In this scenario, one should find a sound argument permitting to choose the proper value of \(a\), and thus the most satisfactory decomposition of \(\langle \vec{S}^2 \rangle\).

### 3. ATOMIC AND DIATOMIC SPIN COMPONENTS

The decomposition of physical quantities into one- and two-center contributions is rooted in the identification of an atom within the molecule. In the 3D-space analysis, the atom is identified by a region of the physical space or atomic domain and its nucleus. Atomic and diatomic contributions can be simply obtained by integration over the corresponding atomic domains \(\Omega_A\) as
\[
\int f(\vec{r}) \, d\vec{r} = \sum_A \int \omega_A(\vec{r}) f(\vec{r}) \, d\vec{r}
\]

\[
\int f(\vec{r}_1, \vec{r}_2) \, d\vec{r}_1 \, d\vec{r}_2 = \sum_{A,B} \int \omega_A(\vec{r}_1) \omega_B(\vec{r}_2) f(\vec{r}_1, \vec{r}_2) \, d\vec{r}_1 \, d\vec{r}_2
\]  

where \(\omega_A\) is a non-negative weight function defined for each atom and each point of the 3D space satisfying \(\sum \omega_A(\vec{r}) = 1\). The one- and two-electron terms of eq 12 can be thus naturally decomposed into one- and two-center contributions, respectively, as shown in eq 13, to obtain
\[
\langle \vec{S}^2 \rangle_A = a \int \omega_A(\vec{r}_1) u(\vec{r}_1) \, d\vec{r}_1 - \frac{1}{2} \int \left[ \Gamma(\vec{r}_1, \vec{r}_2; \vec{r}_1', \vec{r}_2') - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1') \right] \, d\vec{r}_1 \, d\vec{r}_2
\]

\[
\langle \vec{S}^2 \rangle_{AB} = -(1 - 2a) \int \omega_A(\vec{r}_1) \omega_B(\vec{r}_2) \int \left[ \Gamma(\vec{r}_1, \vec{r}_2; \vec{r}_1', \vec{r}_2') - \frac{1}{2} \rho^s(\vec{r}_1; \vec{r}_2) \rho^s(\vec{r}_2; \vec{r}_1') \right] \, d\vec{r}_1 \, d\vec{r}_2
\]  

Equations 14 and 15 can be written more conveniently by using the matrix representations \(D\), \(P\), and \(\Gamma\) of the first-order density matrix, spin-density, and second-order density matrices, respectively, in terms of the molecular orbitals—and the atomic overlap matrix \(S^A\) (see Appendix AII for further details). By using the definition of the effectively unpaired electron density, eq 7, the atomic contribution of the spin squared operator can be written as
\[
\langle \vec{S}^2 \rangle_A = a(2Tr(DS^A) - Tr(D^2S^A)) + \left( \frac{1}{2} - a \right) Tr(P^iS^A P^jS^A)
\]

\[
- (1 - 2a) \sum_{ijkl} \Gamma_{ijkl} S_{kl}^A S_{ij}^A
\]

\[
- \frac{1}{2} \sum_{ijkl} \Gamma_{ijkl} S_{kl}^A S_{ij}^A + \frac{1}{4} Tr(P^iS^A)^2
\]  

Similarly, the diatomic terms can be expressed as follows
\[
\langle \vec{S}^2 \rangle_{AB} = \left( \frac{1}{2} - a \right) Tr(P^iS^A P^jS^B)
\]

\[
- (1 - 2a) \sum_{ijkl} \Gamma_{ijkl} S_{kl}^A S_{ij}^B - \frac{1}{2} \sum_{ijkl} \Gamma_{ijkl} S_{kl}^A S_{ij}^B
\]

\[
+ \frac{1}{4} Tr(P^iS^A) Tr(P^jS^B)
\]  

### 4. SELECTION OF PARAMETER \(a\)

The expressions above illustrate what we referred to in the Introduction as a continuum of closely related decompositions of \(\langle \vec{S}^2 \rangle\) into atomic and diatomic components, which for any value of \(a\) satisfy all of the physical requirements introduced.
For instance, Alcoba et al.’s formulation is based on the expression where \( a = 1/2 \), whereas Mayer and Matito’s original spin-dependent expression for correlated wave functions can be easily rewritten in a form where \( a = 3/8 \) (plus some terms depending on the spin-density that sum up to zero but can lead to changes in the one- and two-center components for non-singlets).

Clearly, the requirements introduced so far for the decomposition of \( \langle \hat{S}^2 \rangle \) are not sufficient. It is necessary to introduce some additional external requirement, based on some physicochemical considerations, which permits one to choose between these possibilities. This can be done by considering how these formulas behave if there is only a single electron in the system.

The value \( \langle \hat{S}^2 \rangle = 3/4 \) is an intrinsic property of the electron, so if one locates an electron in the space with the density (probability density) \( \rho(\vec{r}) \), that means that one locates a spin \( \langle \hat{S}^2 \rangle = 3/4 \) with that probability, too. Therefore, the density of distribution of the \( \langle \hat{S}^2 \rangle \) for a single electron is \( 3/4 \) times the electron density, so that:

\[
\langle \hat{S}^2 \rangle = \frac{3}{4} \int \rho(\vec{r}) \, d\vec{r} \tag{18}
\]

which is the only term on the rhs of eqs 3 and 6 which does not disappear if there is only one electron in the system. Therefore, for a single electron, the atomic components of \( \langle \hat{S}^2 \rangle \) should be equal to the respective gross electron populations multiplied by \( 3/4 \). (No overlap populations may be considered in this case because that would result in a spurious self-coupling of the spin of the given electron with itself.) Now, it is easy to see that in the special case of a single electron, function \( u(\vec{r}) \) equals \( \rho(\vec{r}) \) as a consequence of the idempotency of the first order density matrix. Therefore, eq 12 reproduces eq 18 in the case of a single electron system if, and only if, one chooses \( a = 3/4 \). Interestingly, \( a = 3/4 \) is also the only value for which \( \langle \hat{S}^2 \rangle_\lambda \) is always positive in a general FCI wave function of a minimal-basis-set \( \text{H}_2 \) model (see Appendix AII for further details). Another interesting conclusion drawn from this model system for \( a = 3/4 \) is that one can anticipate small \( \langle \hat{S}^2 \rangle_\lambda \) values at equilibrium distances of singlet molecules, except for true diradicals or systems with strong multireferencial character.

Finally, since for single-determinant wave functions the cumulants are zero by definition, in that case eqs 16 and 17 can be expressed solely in terms of the spin-density matrix elements as

\[
\langle \hat{S}^2 \rangle_\lambda = a \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^A - \hat{P}^A \hat{S}^A \hat{P}^A) + \frac{1}{2} \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^A)^2 + \frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A)^2 \tag{19}
\]

and

\[
\langle \hat{S}^2 \rangle_{\lambda A B} = \left( \frac{1}{2} - a \right) \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^B) + \frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A)^2 \text{Tr}(\hat{P}^B \hat{S}^B) \tag{20}
\]

The original single-determinant formulation of Mayer, also used by Alcoba et al. in ref 8, corresponds to the particular case where \( a = 0 \). It is easy to see that the most recent general formula proposed by Alcoba et al., with \( a = 1/2 \), reduces for the single-determinant case to an expression where for the diatomic terms one has \( \langle \hat{S}^2 \rangle_{\lambda A B} = \langle \hat{S}^2 \rangle_{\lambda A B} \) which is to say, the \( x \) and \( y \) components of the \( \hat{S}^2 \) operator would not contribute to the diatomic spin components.

On the basis of the considerations described above, we suggest the choice \( a = 3/4 \) and hence the following alternative formulation to be used for single-determinant wave functions

\[
\langle \hat{S}^2 \rangle_\lambda = \frac{3}{4} \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^A) - \frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^A)^2 + \frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A)^2 \tag{21}
\]

and

\[
\langle \hat{S}^2 \rangle_{\lambda A B} = -\frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A \hat{P}^B) + \frac{1}{4} \text{Tr}(\hat{P}^A \hat{S}^A) \text{Tr}(\hat{P}^B \hat{S}^B) \tag{22}
\]

It is easy to see that in the case of a single electron or in a ROHF system with a single unpaired electron, eq 22 vanishes, thus avoiding spurious self-coupling of the single electron. In the next section we also provide some numerical evidence supporting the choice \( a = 3/4 \).

### 5. NUMERICAL RESULTS

We have implemented the decompositions of \( \langle \hat{S}^2 \rangle \) described above in the framework of 3D-space analysis for a variety of atomic definitions, namely Becke, Becke-\( \beta \), and Hirshfeld. Table 1. CASSCF(2,4) Atomic \( \langle \hat{S}^2 \rangle_\lambda \) Values for the H\(_2\) Molecule at Optimized Geometries for Several Atomic Definitions and Values of the Parameter \( a \)

<table>
<thead>
<tr>
<th>atom in molecule</th>
<th>( a )</th>
<th>cc-pVDZ</th>
<th>cc-pVTZ</th>
<th>cc-pVQZ</th>
</tr>
</thead>
<tbody>
<tr>
<td>Becke</td>
<td>0</td>
<td>0.109</td>
<td>0.111</td>
<td>0.109</td>
</tr>
<tr>
<td>3/8</td>
<td>0.072</td>
<td>0.073</td>
<td>0.061</td>
<td></td>
</tr>
<tr>
<td>1/2</td>
<td>0.059</td>
<td>0.060</td>
<td>0.034</td>
<td></td>
</tr>
<tr>
<td>3/4</td>
<td>0.034</td>
<td>0.034</td>
<td>0.036</td>
<td></td>
</tr>
<tr>
<td>3/8</td>
<td>0.079</td>
<td>0.080</td>
<td>0.080</td>
<td></td>
</tr>
<tr>
<td>1/2</td>
<td>0.064</td>
<td>0.065</td>
<td>0.066</td>
<td></td>
</tr>
<tr>
<td>3/4</td>
<td>0.035</td>
<td>0.036</td>
<td>0.038</td>
<td></td>
</tr>
<tr>
<td>3/8</td>
<td>0.059</td>
<td>0.060</td>
<td>0.058</td>
<td></td>
</tr>
<tr>
<td>1/2</td>
<td>0.044</td>
<td>0.044</td>
<td>0.045</td>
<td></td>
</tr>
<tr>
<td>3/4</td>
<td>0.028</td>
<td>0.029</td>
<td>0.031</td>
<td></td>
</tr>
</tbody>
</table>

Hirshfeld-iterative and QTAIM (see Appendix AII). The first- and second-order density matrices have been obtained using a modified version of Gaussian 03 and an auxiliary program that reads and processes CISD and CASSCF outputs. All calculations have been carried out at the optimized geometry of the molecule at the current level of theory.

First of all, in Table 1 we compare the values of \( \langle \hat{S}^2 \rangle_\lambda \) calculated for \( \text{H}_2 \) at the equilibrium geometry at the CASSCF(2,4) level for several basis sets and atomic definitions. As expected for a 3D-space analysis, the contributions (in this case, atomic spin-squared values) are rather basis set independent. The effect of the atomic definition on the values of \( \langle \hat{S}^2 \rangle_\lambda \) depends upon the particular choice for the parameter \( a \). In the case of \( a = 0 \), the formula that reduces to the original Mayer’s scheme for the single-determinant case, the values obtained for “fuzzy” atom definitions such as Becke and Hirshfeld are about twice those obtained for QTAIM. The differences decrease for \( a = 3/8 \), which conforms with Mayer’s general formula for singlets, and \( a = 1/2 \), which corresponds to Alcoba’s formulation. In the case of \( a = 3/4 \), the values of \( \langle \hat{S}^2 \rangle_\lambda \) are very small, as

[Note: The rest of the text is not shown due to the nature of the question.]
expected for a diamagnetic molecule, and are very similar for all basis sets and atomic definitions. The Hilbert-space results (not shown) exhibit large basis set dependence. Interestingly, these effects are minimal when using $a = 3/4$, and for that case, the values obtained are very similar to those reported in Table 1.

Table 2. Comparison of the One- And Two-Center Local-Spin Values for the NO Molecule at the UHF/cc-pVTZ Level Using Different Atoms in Molecules Definitions and $a = 3/4$

<table>
<thead>
<tr>
<th>molecule</th>
<th>atom/atom pair</th>
<th>atom</th>
<th>Becke</th>
<th>Becke-$\beta$</th>
<th>Hirshfeld</th>
<th>Hirshfeld iterative</th>
<th>QTAIM</th>
</tr>
</thead>
<tbody>
<tr>
<td>NO</td>
<td>N</td>
<td>0.540</td>
<td>0.497</td>
<td>0.531</td>
<td>0.527</td>
<td>0.547</td>
<td></td>
</tr>
<tr>
<td></td>
<td>O</td>
<td>0.232</td>
<td>0.272</td>
<td>0.241</td>
<td>0.244</td>
<td>0.229</td>
<td></td>
</tr>
<tr>
<td>O, N</td>
<td></td>
<td>0.003</td>
<td>0.002</td>
<td>0.002</td>
<td>0.002</td>
<td>0.005</td>
<td></td>
</tr>
</tbody>
</table>

The implementation and numerical realization in the case of Hilbert-space analysis will be discussed elsewhere.

The effect of the atomic definitions in 3D-space analysis can be more relevant when different types of atoms are present in the molecule, since the relative atomic sizes have a major influence on the values originating from the decomposition of any physical quantity. In Table 2, we report the atomic and diatomic spin contributions obtained for the NO molecule at the UHF/cc-pVTZ level of theory using $a = 3/4$. It can be seen that the differences between the atomic definitions are small and the overall picture is not modified. The molecule is a doublet, and even though the atomic contributions are significant (roughly 0.5 for N and 0.25 for O atoms), the diatomic contribution is essentially zero in all cases. This is a clear indication that there is essentially a single unpaired electron in the system, causing a minor spin-polarization only, and no spurious self-coupling appears. Since the effect of the particular atomic definition is very small, hereafter we report only the values obtained using the Hirshfeld-iterative scheme.

The one- and two-center contributions of $\langle S^2 \rangle$ obtained for a collection of representative singlet systems are gathered in Table 3. The optimized geometries and the wave functions were obtained at the CISD/6-31G** level of theory (six Cartesian $d$ functions). Note that for these systems the overall $\langle S^2 \rangle$ value is zero, but small local atomic spins can be induced by correlation fluctuations. A local-spin analysis is only useful if the genuine antiferromagnetic interactions of localized spins can be clearly distinguished from the electron pairing of $\alpha$ and $\beta$ spins associated with bond formation. All molecules in Table 3 are well described with single-determinant wave functions (except perhaps Li$_2$ and Be$_2$); therefore one should expect very small atomic and diatomic spin contributions. The atomic and diatomic spin values we obtain here are significantly smaller than those reported by Alcoba et al., arising from Hilbert-space analysis. The results indicate that among all values of the parameter $a$, the choice $a = 3/4$ provides the smallest atomic spin values. For instance, in the case of acetylene, a significant, and rather unphysical, $\langle S^2 \rangle_a$ value of 0.339 on the C atom is obtained for $a = 0$, whereas for $a = 3/4$, the value decreases to the much more reasonable 0.09. Interestingly, in the case of the less conventional Li$_2$ and Be$_2$ molecules, the trend is the opposite.

We have also considered open-shell doublet and triplet molecules described at the UHF level, in order to assess the effect of the parameter $a$ on the single-determinant description. The one- and two-center contributions of $\langle S^2 \rangle$ using Hirshfeld-iterative atomic definitions are collected on Tables 4 and 5, respectively. The optimized geometries and wave functions were obtained at the UHF/cc-pVTZ level of theory. The overall $\langle S^2 \rangle$ value is also reported, since significant spin contamination is present in some cases like the CN or CH radicals. Note that the case of $a = 0$ corresponds to the original formulation of Mayer for single-determinant wave functions, whereas $a = 1/2$ would correspond to the “spin-free” formulation introduced by Alcoba et al.9

In the case of doublet systems, the differences in the atomic spin contributions for different values of the parameter $a$ are smaller than for the singlets. The sign of diatomic contributions often changes going from $a = 0$ to the preferred $a = 3/4$ value. However, except in the case of NO and CN radicals, the spin is essentially localized on one atom, and therefore the diatomic
contributions are very small. Interestingly, the atomic spin contribution on the H atoms for radicals like OH, CH, CH₃, or NH₂ is essentially zero when \( a = 0 \), yet there is significant (between 0.022 and 0.072) diatomic contribution involving the H atom. This rather puzzling picture changes with \( a = 3/4 \), where the local spin on the H atom increases but the diatomic contribution between the atoms is essentially zero, again in accordance with the model of a single unpaired electron which is significantly delocalized. In the case of CH and CN radicals, there is a significant spin contamination that results in atomic spin contributions larger than the value of 3/4 of a single localized electron. In both cases, the local spin on the C atoms is larger than 1, in line with overall \( \langle S^2 \rangle \) values of 1.106 and 1.075, respectively. The diatomic contribution in the case of CN is significant and negative (up to \(-0.23\)), indicating some degree of antiferromagnetic coupling between the centers. However, it is difficult to draw conclusions from a wave function with such an amount of spin contamination.

The results for a set of molecules in their lowest triplet state are collected in Table 5. The UHF wave function exhibits small spin contamination for all molecules. In contrast to the rather small differences found for the doublets, the local-spin contributions depend considerably on the value of the parameter \( a \) in the case of homonuclear diatomic molecules, especially for Li₂ and O₂. Note that for \( a = 3/4 \) the atomic spin contribution on Li and O atoms is larger than the value of a localized electron (3/4), whereas for \( a = 1/2 \) the atomic spin is very close to that number. In fact, it can be easily seen that in the case of Li₂ described at the ROHF level of theory, the value \( a = 1/2 \) yields exactly \( \langle S^2 \rangle_A = 3/4 \) and \( \langle S^2 \rangle_{AB} = 1/4 \) at any interatomic distance and for any atomic definition.

![Figure 1](link)

**Table 4. One- and Two-Center Local-Spin Values for Doublet Systems at the UHF/cc-pVTZ Level for Different Values of the Parameter \( a \)**

<table>
<thead>
<tr>
<th>molecule</th>
<th>atom/atom pair</th>
<th>( \langle S^2 \rangle )</th>
<th>( a = 0 )</th>
<th>( a = 3/8 )</th>
<th>( a = 1/2 )</th>
<th>( a = 3/4 )</th>
</tr>
</thead>
<tbody>
<tr>
<td>OH</td>
<td>O</td>
<td>0.756</td>
<td>0.712</td>
<td>0.726</td>
<td>0.731</td>
<td>0.740</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.001</td>
<td>0.015</td>
<td>0.019</td>
<td>0.028</td>
<td></td>
</tr>
<tr>
<td>NO</td>
<td>N</td>
<td>0.767</td>
<td>0.475</td>
<td>0.449</td>
<td>0.370</td>
<td>0.317</td>
</tr>
<tr>
<td></td>
<td>O</td>
<td>0.087</td>
<td>0.166</td>
<td>0.192</td>
<td>0.244</td>
<td></td>
</tr>
<tr>
<td>CN</td>
<td>C</td>
<td>1.073</td>
<td>1.084</td>
<td>1.147</td>
<td>1.168</td>
<td>1.211</td>
</tr>
<tr>
<td></td>
<td>N</td>
<td>0.199</td>
<td>0.262</td>
<td>0.283</td>
<td>0.325</td>
<td></td>
</tr>
<tr>
<td>NH₂</td>
<td>N</td>
<td>0.759</td>
<td>0.649</td>
<td>0.681</td>
<td>0.691</td>
<td>0.712</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.001</td>
<td>0.018</td>
<td>0.024</td>
<td>0.035</td>
<td></td>
</tr>
<tr>
<td></td>
<td>O</td>
<td>0.026</td>
<td>0.010</td>
<td>0.005</td>
<td>0.005</td>
<td></td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.001</td>
<td>0.000</td>
<td>0.000</td>
<td>0.000</td>
<td></td>
</tr>
<tr>
<td>CH</td>
<td>C</td>
<td>1.106</td>
<td>0.955</td>
<td>1.000</td>
<td>1.015</td>
<td>1.045</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.007</td>
<td>0.052</td>
<td>0.067</td>
<td>0.098</td>
<td></td>
</tr>
<tr>
<td>CH₃</td>
<td>C</td>
<td>0.761</td>
<td>0.544</td>
<td>0.601</td>
<td>0.620</td>
<td>0.658</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.003</td>
<td>0.025</td>
<td>0.032</td>
<td>0.046</td>
<td></td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.002</td>
<td>0.001</td>
<td>0.000</td>
<td>0.000</td>
<td></td>
</tr>
</tbody>
</table>

**Table 5. One- and Two-Center Local-Spin Values for Triplet Systems at the UHF/cc-pVTZ Level for Different Values of the Parameter \( a \)**

<table>
<thead>
<tr>
<th>molecule</th>
<th>atom/atom pair</th>
<th>( \langle S^2 \rangle )</th>
<th>( a = 0 )</th>
<th>( a = 3/8 )</th>
<th>( a = 1/2 )</th>
<th>( a = 3/4 )</th>
</tr>
</thead>
<tbody>
<tr>
<td>Li₂</td>
<td>Li</td>
<td>1.988</td>
<td>0.497</td>
<td>0.683</td>
<td>0.745</td>
<td>0.870</td>
</tr>
<tr>
<td></td>
<td>Li, Li</td>
<td>0.497</td>
<td>0.311</td>
<td>0.248</td>
<td>0.124</td>
<td></td>
</tr>
<tr>
<td>O₂</td>
<td>O</td>
<td>2.042</td>
<td>0.540</td>
<td>0.713</td>
<td>0.770</td>
<td>0.886</td>
</tr>
<tr>
<td></td>
<td>O, O</td>
<td>0.481</td>
<td>0.308</td>
<td>0.250</td>
<td>0.135</td>
<td></td>
</tr>
<tr>
<td>NF</td>
<td>N</td>
<td>2.024</td>
<td>1.607</td>
<td>1.681</td>
<td>1.706</td>
<td>1.756</td>
</tr>
<tr>
<td></td>
<td>F</td>
<td>0.030</td>
<td>0.104</td>
<td>0.129</td>
<td>0.179</td>
<td></td>
</tr>
<tr>
<td></td>
<td>N, F</td>
<td>0.193</td>
<td>0.119</td>
<td>0.094</td>
<td>0.044</td>
<td></td>
</tr>
<tr>
<td>NH</td>
<td>N</td>
<td>2.015</td>
<td>1.846</td>
<td>1.885</td>
<td>1.898</td>
<td>1.925</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.005</td>
<td>0.044</td>
<td>0.057</td>
<td>0.084</td>
<td></td>
</tr>
<tr>
<td></td>
<td>N, H</td>
<td>0.082</td>
<td>0.043</td>
<td>0.030</td>
<td>0.003</td>
<td></td>
</tr>
<tr>
<td>C₂</td>
<td>C</td>
<td>2.009</td>
<td>0.677</td>
<td>0.735</td>
<td>0.754</td>
<td>0.793</td>
</tr>
<tr>
<td></td>
<td>C, C</td>
<td>0.327</td>
<td>0.269</td>
<td>0.250</td>
<td>0.211</td>
<td></td>
</tr>
<tr>
<td>CH₂</td>
<td>C</td>
<td>2.016</td>
<td>1.612</td>
<td>1.696</td>
<td>1.724</td>
<td>1.780</td>
</tr>
<tr>
<td></td>
<td>H</td>
<td>0.006</td>
<td>0.052</td>
<td>0.067</td>
<td>0.097</td>
<td></td>
</tr>
<tr>
<td></td>
<td>C, H</td>
<td>0.095</td>
<td>0.053</td>
<td>0.039</td>
<td>0.010</td>
<td></td>
</tr>
<tr>
<td></td>
<td>H, H</td>
<td>0.006</td>
<td>0.003</td>
<td>0.002</td>
<td>0.000</td>
<td></td>
</tr>
</tbody>
</table>
other hand, the results for the NF, NH, and CH₄ molecules are consistent with the presence of two unpaired electrons located on the N and C atoms, respectively, with negligible diatomic contributions.

Finally, Figure 1 displays the \( \langle \hat{S}^2 \rangle \) values as a function of the parameter \( a \) along the dissociation curve of several homonuclear diatomic molecules, namely, H₂, N₂, and O₂, in their lowest-lying singlet state described with a proper full-valence CASSCF wave function. As stated above, at the dissociation limit, the atomic spin values tend to the value of the lowest-lying singlet state described with a proper full-valence determinant and correlated wave functions has been intro-

\[ \langle \hat{S}^2 \rangle_a = \sum_{i=1}^{N} \frac{\alpha_i^2}{2} \sum_{\sigma} \left| \langle \sigma \rangle \right|^2 \]

where \( \sigma \) is the spin quantum number. In the case of \( N \), at very short interatomic distances (ca. 0.8 Å), a small correspondingly free atoms, i.e., 3/4, 15/4, and 2 in the case of H₂, N₂, and O₂, respectively, for all values of \( a \). In the case of N₂, a parameter \( a \) can be attributed to the integration error associated with any 3D-space analysis and is not considered to be of any special relevance.

6. CONCLUSIONS

We have shown that there is a continuum of different formulations for the decomposition of \( \langle \hat{S}^2 \rangle \) that fulfills all physical requirements imposed to date. We have presented a new criterion based upon the behavior of single-electron systems and a simple two-electron model by which the value of the parameter defining that continuum can be properly fixed, and based on this, a new general formula applicable for both single-determinant and correlated wave functions has been introduced. The scheme has been realized in the three-dimensional physical space and implemented for Bader’s QTAIM and a number of “fuzzy” atom definitions. We provide numerical data for a set of representative closed-shell and open-shell systems and for the dissociation curves of several diatomic molecules. The results have been compared with other existing formulations. Contrary to the most recent results, very small local-spin values are obtained for genuine diamagnetic molecules treated at the correlated level, in conformity with the physical expectations.

APPENDIX

A. Expectation Value of the \( \hat{S}^2 \) Operator

The \( \hat{S}^2 \) operator can be written as a sum of one- and two-electron operators acting on the spin coordinates of an N-electron wave function

\[ \hat{S}^2 = \sum_{i} \left[ S_{x}^2(i) + S_{y}^2(i) + S_{z}^2(i) \right] + \sum_{i,j} \left[ S_{x}(i)S_{x}(j) + S_{y}(i)S_{y}(j) + S_{z}(i)S_{z}(j) \right] \]

The action of the different individual one-electron spin operators over the corresponding \( \alpha \) and \( \beta \) one-electron spin functions yields the well-known relations (in a.u.)

\[ \hat{S}_x(\alpha) = \frac{1}{2}(\beta) \quad \hat{S}_y(\alpha) = i \beta \quad \hat{S}_z(\alpha) = \frac{1}{2}(\alpha) \]

\[ \hat{S}_x(\beta) = \frac{1}{2}(\alpha) \quad \hat{S}_y(\beta) = -i \alpha \quad \hat{S}_z(\beta) = \frac{1}{2}(\beta) \]

The expectation value of a symmetric sum of Hermitian one-electron operators of the form \( \sum_i h(i) \) for an antisymmetric wave function \( |\Psi\rangle \) can be written in terms of the corresponding one-electron reduced density matrix (1-RDM). In the case of the one-electron part of eq 23, one can make use of the appropriate relations of eq 24 and integrate over the spin coordinates to obtain

\[ \langle \Psi | \sum_{i,j} \hat{S}_x(i) \hat{S}_z(j) |\Psi\rangle = \frac{\alpha_1 \beta_1}{2} \int \left[ \hat{S}_x^2(\alpha_1) + \hat{S}_y^2(\alpha_1) + \hat{S}_z^2(\beta_1) \right] \rho_\sigma(\tau_1; \tau_2) d\tau_1 d\tau_2 \]

where \( \tau = (\tau_x, \tau_y, \tau_z) \) stands for the spatial and spin coordinates of the electron.

In the case of a sum of two-electron operators, the expectation value can be expressed quite analogously through the second-order reduced density matrix (2-RDM)

\[ \langle \Psi | \sum_{i,j} \hat{S}_x(i) \hat{S}_z(j) |\Psi\rangle = \frac{\alpha_1 \beta_1}{2} \int \left[ \hat{S}_x^2(\alpha_1) + \hat{S}_y^2(\alpha_1) + \hat{S}_z^2(\beta_1) \right] \rho_\sigma(\tau_1; \tau_2) \rho_\sigma(\tau_1; \tau_2) d\tau_1 d\tau_2 \]

As for the terms involving either \( \hat{S}_x \) or \( \hat{S}_z \)-type operators, each such operator flips the spin function on which it acts; therefore, only those terms of the 2-RDM do not vanish upon
integration over the spin coordinates, for which \( \sigma_i \neq \sigma'_i \) and \( \sigma_j \neq \sigma'_j \):

\[
\langle \Psi | \sum_{i \neq j} \hat{\rho}_x(i) \hat{\rho}_x(j) | \Psi \rangle
\]

\[
= \sum_{\sigma_1, \sigma'_1, \sigma_2, \sigma'_2} \int \hat{\rho}_x(s_1) \hat{\rho}_x(s_2) \\
\times \rho_2^{\sigma_1 \sigma_2} \sigma_1 \sigma_2 (r'_1, r'_2; r_1, r_2) \left| \begin{array}{c} x'_1 = \bar{x}_1 \\ x'_2 = \bar{x}_2 \end{array} \right|
\]

\[
= \frac{1}{4} \int \rho_2^{a\alpha\beta\alpha\beta} (r_1, r_2; r_1, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{4} \int \rho_2^{a\alpha\beta\beta\alpha} (r_1, r_2; r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
- \frac{1}{4} \int \rho_2^{a\beta\alpha\alpha\beta} (r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2 \\
- \frac{1}{4} \int \rho_2^{a\beta\beta\alpha\alpha} (r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{2} \int \rho_2^{a\beta\alpha\beta\alpha} (r_1, r_2; r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{2} \int \rho_2^{a\beta\beta\beta\alpha\alpha} (r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2
\]

(28)

It is easy to see that exactly the same terms originate from the contribution of the \( \hat{S}_z \)-type operator.

Adding up eqs 25, 27, and twice 28, one obtains

\[
\langle S^2 \rangle = \frac{3}{4} \int \rho(r_1; r_1) \, d\bar{r}_1 + \frac{1}{4} \int \rho^{\alpha\alpha\alpha\alpha}(r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{4} \int \rho^{\alpha\alpha\beta\beta}(r_1, r_2; r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
- \frac{1}{4} \int \rho^{\alpha\beta\alpha\beta}(r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2 \\
- \frac{1}{4} \int \rho^{\alpha\beta\beta\alpha}(r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{2} \int \rho^{\alpha\beta\beta\beta}(r_1, r_2; r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
+ \frac{1}{2} \int \rho^{\alpha\beta\beta\beta\alpha}(r_1, r_2; r_2, r_1) \, d\bar{r}_1 \, d\bar{r}_2
\]

(29)

Moreover, since the 2-RDM satisfies

\[
\rho_2^{\sigma_1 \sigma_2 \sigma_1 \sigma_2}(r_1, r_2; r_2, r_1) = -\rho_2^{\sigma_1 \sigma_2 \sigma_1 \sigma_2}(r_1, r_2; r_2, r_1)
\]

(30)

one can add and subtract the terms \((1/2) \int \rho^{\alpha\alpha\alpha\alpha}(r_1, r_3, r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_3 \, d\bar{r}_2\) and \((1/2) \int \rho^{\alpha\alpha\beta\beta}(r_1, r_3, r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_3 \, d\bar{r}_2\) to eq 29 and make use of eq 30 to obtain a final formula for \(\langle S^2 \rangle\) conveniently expressed in terms of the spinless first- and second-order density matrices as

\[
\langle S^2 \rangle = \frac{3}{4} \int \rho(r_1; r_1) \, d\bar{r}_1 - \frac{1}{4} \int \rho_2(r_1, r_2; r_1, r_2) \, d\bar{r}_1 \, d\bar{r}_2 \\
- \frac{1}{2} \int \rho_2(r_1, r_2; r_2, r_2) \, d\bar{r}_1 \, d\bar{r}_2
\]

(31)

where

\[
\rho_2(r_1, r_2; r_2, r_1) = \rho_2^{\alpha\alpha\alpha\alpha}(r_1, r_2; r_1, r_2) \\
+ \rho_2^{\alpha\alpha\beta\beta}(r_1, r_2; r_1, r_2) \\
+ \rho_2^{\alpha\beta\alpha\beta}(r_1, r_2; r_1, r_2) \\
+ \rho_2^{\alpha\beta\beta\alpha}(r_1, r_2; r_1, r_2)
\]

(32)

All. 3D-Space Analysis

In the classical Hilbert-space atomic partition, each atom is identified by its nucleus and the subspace of the basis atomic orbitals centered on it. In the 3D-space analysis, the atom is identified by a region of the physical space or atomic domain (and its nucleus, of course). If the atomic domains are disjoint, then the decomposition of a physical quantity into atomic contributions can be simply carried out by integration over the corresponding atomic domains \(\Omega_A\) as

\[
\int f(\bar{r}) \, d\bar{r} = \sum_A \int_{\Omega_A} f(\bar{r}) \, d\bar{r}
\]

\[
\int f(\bar{r}_1, \bar{r}_2) \, d\bar{r}_1 \, d\bar{r}_2 = \sum_{A,B} \int_{\Omega_A} \int_{\Omega_B} f(\bar{r}_1, \bar{r}_2) \, d\bar{r}_1 \, d\bar{r}_2
\]

(33)

where one can see that one- and two-electron quantities quite naturally decompose into one-center and one- and two-center contributions, respectively.

One may, however, consider also so-called “fuzzy” atomic domains, having no strict boundaries. In order to treat the most different schemes in a common framework, one can introduce a non-negative weight function \(w_A(\bar{r})\) for each atom and each point of the 3D space satisfying the requirement

\[
\sum_A w_A(\bar{r}) = 1
\]

(34)

In the case of the “fuzzy” atoms, the value of \(w_A(\bar{r})\) is large in the vicinity of the nucleus of atom \(A\) and quickly becomes negligible outside. In the special case of disjoint atomic domains, such as those of Bader’s Atoms in Molecules theory,\(^{15}\) \(w_A(\bar{r}) = 1\) for points inside the atomic domain of \(A\) and \(w_A(\bar{r}) = 0\) outside of it.

Then, the decomposition of a physical quantity into atomic contributions can be performed by inserting the identity 34 one or two times for the one- and two-electron integrals, respectively

\[
\int f(\bar{r}) \, d\bar{r} = \sum_A \int w_A(\bar{r}) f(\bar{r}) \, d\bar{r}
\]

\[
\int f(\bar{r}_1, \bar{r}_2) \, d\bar{r}_1 \, d\bar{r}_2 = \sum_{A,B} \int_{\Omega_A} \int_{\Omega_B} w_A(\bar{r}_1) w_B(\bar{r}_2) f(\bar{r}_1, \bar{r}_2) \, d\bar{r}_1 \, d\bar{r}_2
\]

(35)

In the past, we have made use of the simplest Becke’s atoms\(^{16}\) to show how several quantities such as bond orders, overlap populations, or energy components can be obtained for “fuzzy” atoms.\(^{25}\) The shape of such Becke atoms is determined by a “stiffness parameter” and by a set of fixed atomic radii that define the relative size of the atomic basins. One can also use the position of the minima of the total density along the internuclear axis\(^{26}\) for each pair of atoms to determine the ratio of the atomic radii. Such a scheme, referred to as Becke-\(\rho\) in ref 20 can be considered as a good and efficient adaptation of some ideas of Bader’s theory.

Hirshfeld\(^{21}\) defined another way to partition the molecular space using promolecular densities. Lately, the iterative-Hirshfeld approach,\(^{22}\) improving over classical Hirshfeld’s, has been suggested. In both schemes, the atomic weight of atom \(A\) at a given point in the space is determined by the ratio

\[
w_A(\bar{r}) = \frac{\rho_A(\bar{r})}{\sum_B \rho_B(\bar{r})}
\]

(36)
where \( \rho^A_0(\vec{r}) \) represents a promolecular density of the atom \( A \).

In the classical Hirshfeld, the resulting shape of the atoms in the molecule is strongly dependent on the choice of the promolecular state of the atom. The improved Hirshfeld-iterative scheme corrects this problem through an iterative process to obtain promolecular atomic densities that integrate to the same (usually fractional) number of electrons as do the atoms in the molecule.

We have made use of all “fuzzy” atom definitions mentioned above, as well as Bader’s atomic domains, to obtain the one- and two-center contributions of \( \langle \hat{S}^2 \rangle \). In this paper, we show that different atomic definitions induce only minor qualitative differences in the results.

Instead of using the formulae 35 to the first order density matrix, the spin-density matrix, or the cumulant, one can express them in terms of the molecular (or natural) orbitals as

\[
\rho(\vec{r}, \vec{r}’) = \sum_{ij} D_{ij} \phi_i(\vec{r}) \phi_j(\vec{r}’) 
\]

(37)

\[
\rho^s(\vec{r}, \vec{r}’) = \sum_{ij} P_{ij}^s \phi_i^s(\vec{r}) \phi_j^s(\vec{r}’) 
\]

(38)

\[
\Gamma^2(\vec{r}_1, \vec{r}_2; \vec{r}_1’, \vec{r}_2’) = \sum_{ijkl} \Gamma_{ijkl}^s \phi_i(\vec{r}_1) \phi_j^*(\vec{r}_1’) \phi_k(\vec{r}_2) \phi_l^*(\vec{r}_2’) 
\]

(39)

and reduce the integrations to calculating the atomic overlap matrix

\[
S_{ij}^A = \int \psi_i^*(\vec{r}) \psi_j(\vec{r}) \, d\vec{r} 
\]

(40)
in the basis of molecular orbitals.

**All. The \( \text{H}_2 \) Model**

A highly desirable requirement for the decomposition scheme of \( \langle \hat{S}^2 \rangle \) is that the one-center contributions should be non-negative. Let us consider the \( \text{H}_2 \) molecule with minimal basis at the full-CI level. The molecular orbitals are determined from symmetry considerations, and the ground state wave function can be written as \( \Psi = (1 - c_1^2)^{1/2} \Psi_0 + c_1 \Psi_1 \), where \( \Psi_0 \) and \( \Psi_1 \) are the ground-state and double-excited Slater determinants, respectively. For this model system, taking \( S_{12} = S_{21} = 1/2 \) for the atomic overlap matrix element (the exact value at the dissociation limit), eq 16 can be expressed as

\[
\langle \hat{S}^2 \rangle_A = 4ac_1^2(1 - c_1^2) + \left( \frac{3}{2} - 2a \right) c_1 \sqrt{1 - c_1^2}
\]

(41)

In Figure 2, we plot \( \langle \hat{S}^2 \rangle_A \) vs \( c_1 \) for several values of the parameter \( a \). At the dissociation limit of the ground state, \( (c_1 = -(1/2)^{1/2}) \), all curves tend to the proper value \( \langle \hat{S}^2 \rangle_A = 3/4 \). Similarly, all curves predict zero local spin for Hartree-Fock wave functions \( (c_1 = 0) \). However, only for the special case of \( a = 3/4 \) are all values of \( \langle \hat{S}^2 \rangle_A \) positive for any value of the mixing coefficient \( c_1 \).

Another interesting property in the case of \( a = 3/4 \) is that the second term in eq 41—which is linear in the coefficient \( c_1 \) as \( c_1 \) tends to zero—vanishes. Note that this is the leading term in singlet systems that are well-described with a single closed-shell determinant wave function, as the first term on the rhs would be quadratic in \( c_1 \). This shows that very small local-spin values are thus obtained for genuine diamagnetic molecules, if the preferred value \( a = 3/4 \) is used, in accordance with the physical expectations.

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**Notes**

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